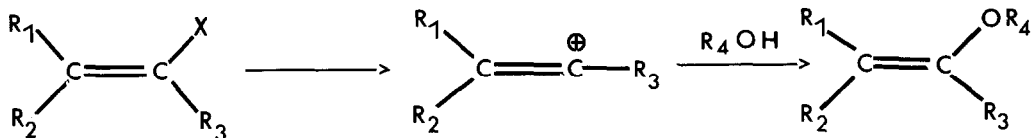


SOLVOLYSIS AND REARRANGEMENT OF ARYL VINYL COMPOUNDS
IN PYRIDINE HYDROCHLORIDE

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Summary Anisyl substituted stilbenes undergo concomitant thermal rearrangement and loss of a halogen atom with dealkylation in pyridine hydrochloride.

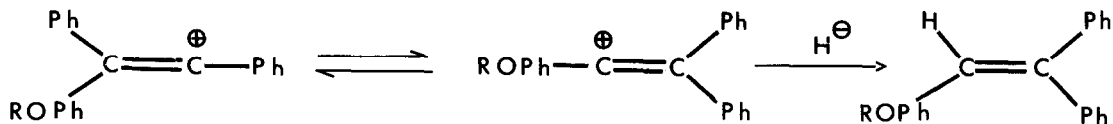
The reaction of aryl substituted vinyl halides in strongly ionizing solvents is thought to proceed via a vinyl cation intermediate (see review¹), which subsequently reacts with solvent (or eliminates if β hydrogens are present) (Scheme I).



SCHEME I

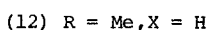
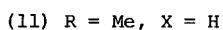
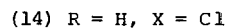
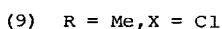
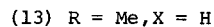
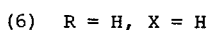
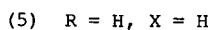
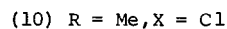
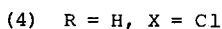
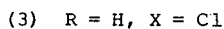
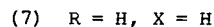
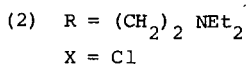
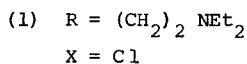
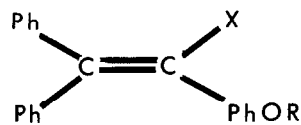
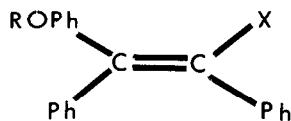
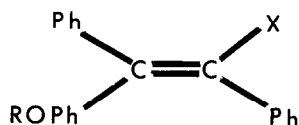
While attempting the dealkylation of clomiphene² (1) and (2) to phenols (3) and (4) in search of a sensitive assay for clomiphene, based on the fluorescence of the dansyl derivatives of phenols (3) and (4), a mixture of three isomeric phenols (5), (6) and (7) (approx. 1:1:1) was isolated³. The proposed structures of these three isomeric phenols were confirmed by their individual syntheses^{4,5} and co-injection of their TMS ethers on GC with the silylated mixture obtained from the reaction of clomiphene⁶.

The phenols (5), (6) and (7) may be rationalized to form by a reversible 1,2 aryl shift⁷ within the initially formed vinyl cation followed by hydride abstraction, possibly from the dealkylated side chain (Scheme II).



SCHEME II

Extremely vigorous conditions are necessary to produce the vinyl cation since demethylation of ethers (8), (9) and (10) with pyridine hydrochloride (200°, 2h), employed to prepare phenols (3), (4) and (14) caused no detectable loss of chlorine, whereas a mixture of ethers (8) and (9) afforded phenols (5), (6) and (7) (1:1:1) under the conditions used for clomiphene⁸.



The reaction was not limited to vinyl chlorides for, separately, ether (13) and a mixture of ethers (11) and (12) afforded phenols (5), (6) and (7) (1:1:1) when heated with pyridine hydrochloride at 230°.

The authors thank Dr. W.L. Albrecht of Merrell - National Laboratory for a generous gift of clomiphene citrate and Mr. C.J. Seaborn for technical assistance.

References and Footnotes

1. P.J. Stang, Prog. Phys. Org. Chem., 10, 257 (1973).
2. Clomiphene, a 1:1 mixture of cis and trans 2- (p - (2-Chloro-1,2-diphenyl-vinyl) phenoxy) triethylamine, is marketed as the citrate salt.
3. Clomiphene citrate (100 mg) and pyridine hydrochloride (2 g) were heated at 230° for 24 h
4. E.C. Dodds, L. Goldberg, E.I. Grunfeld, W. Larson, C.M. Saffer, Jnr, and R. Robinson, Proc. Roy. Soc., 132B, 83 (1944).
5. The p-toluenesulphonate of phenol (7), m.p. 116-117°, gave a satisfactory combustion analysis.
6. SE-30 SCOT column, 20 m x 0.5 mm ID, T = 240°, Neff = 15,000; Retention indices for the TMS derivatives of (5), (7) and (6) were 2467, 2495 and 2510 respectively; Resolution of (7) and (6) was 0.48.
7. D.Y. Curtin, J.H. Kampmeier and M.L. Farmer, J. Amer. Chem. Soc., 87, 863 (1965).

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